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"Assistant Commissioner for Patents Washington, D.C. 20231"

on July 12, 2001

RIMMA MITELMAN

Class Mail in an envelope addressed to:

Date of Signature

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IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

Applicant:

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Serial No.:

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For:

DETERGENT COMPOSITION COMPRISING BENEFIT AGENTS

Edgewater, New Jersey 07020

July 12, 2001

SUBMISSION OF PRIORITY DOCUMENT

Assistant Commissioner for Patents Washington, D.C. 20231

Sir:

Pursuant to rule 55(b) of the Rules of Practice in Patent Cases, Applicant(s) is/are submitting herewith a certified copy of the European Application No. 99310428.0 filed December 22, 1999, upon which the claim for priority under 35 U.S.C. § 119 was made in the United States.

It is respectfully requested that the priority document be made part of the file history.

Respectfully submitted,

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Attestation

Die angehefteten Unterlagen stimmen mit der ursprünglich eingereichten Fassung der auf dem nächsten Blatt bezeichneten europäischen Patentanmeldung überein.

The attached documents are exact copies of the European patent application conformes à la version described on the following page, as originally filed.

Les documents fixés à cette attestation sont initialement déposée de la demande de brevet européen spécifiée à la page suivante.

Patentanmeldung Nr.

Patent application No. Demande de brevet n°

99310428.0

Der Präsident des Europäischen Patentamts; Im Auftrag

For the President of the European Patent Office

Le Président de l'Office européen des brevets p.o.

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Blatt 2 der Bescheinigung Sheet 2 of the certificate Page 2 de l'attestation

Anmeldung Nr.:

Application no.: Demande n°:

99310428.0

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Applicant(s): Demandeur(s): UNILEVER PLC

London EC4P: 4BQ UNITED KINGDOM

Bezeichnung der Erfindung: Title of the invention: Titre de l'invention:

Detergent compositions comprising benefit agents

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DETERGENT COMPOSITIONS COMPRISING BENEFIT AGENTS

TECHNICAL FIELD

The present invention generally relates to detergent compositions comprising benefit agents. More in particular, it relates to detergent compositions comprising fusion proteins which are useful for delivering a benefit agent to a fabric.

10 BACKGROUND AND PRIOR ART

WO-A-98/00500 (Unilever) discloses detergent compositions wherein a Benefit Agent is delivered onto fabric by means of peptide or protein Deposition Aid having a high affinity for fabric. The Benefit Agent can be a fabric softening agent, a perfume, a polymeric lubricant, a photoprotective agent, a latex, a resin, a dye fixative agent, an encapsulated material, an antioxidant, an insecticide, a soil repelling agent or a soil release agent. The Benefit Agent is attached or adsorbed to a peptide or protein Deposition Aid having a high affinity for fabric. Preferably, the deposition aid is the cellulose binding domain of a cellulase enzyme. The compositions are said to effectively deposit the Benefit Agent onto the fabric during the wash cycle.

- The detergent compositions disclosed in WO-A-98/00500 have a number of drawbacks. The Benefit Agent has to be attached/adsorbed to the Deposition Aid. If it is to be attached, it is preferably done by means of a linking agent. However, not all benefit agents can be derivatised.
- 30 Furthermore, linking the Benefit Agent to the Deposition Aid may negatively affect the properties of the Benefit Agent. For example, perfumes cannot be covalently bound to larger molecules without negatively affecting their perfume properties.

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It is therefore an object of the present invention to provide an alternative or improved detergent composition which is capable of delivering a benefit agent to a fabric during a washing or rinsing process.

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Surprisingly, we have now found that these and other objects of the invention may be achieved by the detergent compositions of the present invention, which are characterised in that they comprise a fusion protein comprising a cellulose binding domain and a domain having a high binding affinity for another ligand. This fusion molecule is bifunctional in its binding ability, whereby the cellulose binding domain region binds to cellulosic based materials and the second domain binds to another ligand.

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According to DE-A-196 21 224 (Henkel), the transfer of textile dyes from one garment to another during a washing or rinsing process may be inhibited by adding antibodies against the textile dye to the wash or rinse liquid.

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WO-A-98/07820 (P&G) discloses amongst others rinse treatment compositions containing antibodies directed at cellulase and standard softener actives (such as DEQA).

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DEFINITION OF THE INVENTION

According to a first aspect of the invention, there is provided a fusion protein comprising a cellulose binding domain and a domain having a high binding affinity for another ligand. Preferably, the domain having a high binding affinity is directed at a Benefit Agent.

According to a second aspect, there is provided a detergent composition comprising said fusion protein.

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DETAILED DESCRIPTION OF THE INVENTION

1.1 The cellulose binding domain

In its first aspect, the invention relates to a fusion protein comprising a cellulose binding domain (CBD) and a domain having a high binding affinity for another ligand. The cellulose binding domain is a part of most cellulase enzymes and can be obtained therefrom. Cellulose binding domains are also obtainable from xylanase and other hemicellulase degrading enzymes. Preferably, the cellulose binding domain is obtainable from a fungal enzyme origin such as Humicola, Trichoderma, Thermomonospora, Phanerochaete, Aspergillus or from a bacterial enzyme origin such as <a href="https://docume.com/Bacterial-enzyme.com/Ba

In the fusion protein according to the invention, the cellulose binding domain is fused to a second domain having a high binding affinity for another ligand. Preferably, the cellulose binding domain is connected to the domain having a high binding affinity for another ligand by means of a linker consisting of about 0-20, preferably about 2-15, more preferably of 2-5 amino acids.

- The second domain having a high binding affinity for another ligand may, for instance, be an antibody or an antibody fragment. Especially preferred are heavy chain antibodies such as found in Camelidae.
- 30 Generally speaking, the degree of binding of a molecule A to another molecule B can be generally expressed by the chemical equilibrium constant K_d resulting from the following reaction:

 $[A] + [B] \Leftrightarrow [A = B]$

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The chemical equilibrium constant Ka is then given by:

$$K_d = \frac{[A] \times [B]}{[A = B]}$$

Whether the binding of a molecule to the fabric / ligand is specific or not can be judged from the difference between the binding (Kd value) of the molecule to one type of fabric 5 / ligand, versus the binding to another type of fabric / ligand material. For applications in laundry, said ligand material will form part of or be associated with a benefit agent. In this aspect of the invention, the CBD region of the fusion protein binds to the fabric and the high affinity 10 domain region binds to the benefit agent. Alternatively this approach can be reversed whereby the fabric material, or a ligand bound to the fabric is targeted by the high affinity domain of the fusion protein and the cellulose binding domain region binds to a cellulosic based or cellulosic 15 containing benefit agent. However, it will usually be more convenient to measure Kd values and differences in Kd values on other materials such as a polystyrene microtitre plate or a specialised surface in an analytical biosensor. The difference between the two binding constants should be 20 minimally 10, preferably more than 100, and more preferably, more that 1000. Typically, the reagent should bind to the ligand / fabric, with a Ka lower than 10-4 M, preferably lower than 10-6 M and could be 10-10 M or even less. Higher binding affinities (Kd of less than 10°5 M) and/or a larger 25 difference between the one type of ligand / fabric and another type of ligand / fabric (or background) binding would increase the deposition of the benefit agent. Also, the weight efficiency of the reagent in the total rinse composition would be increased and smaller amounts of the 30 reagent would be required.

Several classes of reagent or molecules can be envisaged which deliver the capability of specific binding to fabrics

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/ ligands, to which one would like to deliver the benefit agent. In the following we will give a number of examples of such molecules having such capabilities, without pretending to be exhaustive.

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1.2.1. Antibodies.

Antibodies are specific binding proteins. Their function in nature is to protect against disease by recognising (and binding) foreign bodies, such as viruses or Bacteria, but not self-cells. Furthermore, methods are well-known in the art to generate antibodies that are specific for almost any protein, organic molecule, or cell surface, that is likely to be encountered. This binding specificity has been exploited in the Biotechnology industry, principally for 15 medical diagnostics. For example, many home-based pregnancy test kits comprise an antibody that specifically binds to the pregnancy marker hormone, human chorionic gonadotrop1n (hCG), but not to other hormones present in urine.

20 More recently, the use of antibodies in laundry products has been described (Henkel, Procter and Gamble, Unilever). In particular, Unilever has described the use of stain-specific antibodies to target bleaching enzymes exclusively to stains but not to dyes - thus achieving efficient stain removal 25 without damaging surrounding fabric.

Antibodies are well known examples of molecules which are capable of binding specifically to compounds against which they were raised. Antibodies can be derived from several 30 sources. From mice, monoclonal antibodies can be obtained which possess very high binding affinities. From such antibodies, Fab, Fv or scFv fragments, can be prepared which have retained their binding properties. Such antibodies or fragments can be produced through recombinant DNA technology 35 by microbial fermentation. Well known production hosts for

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antibodies and their fragments are yeast, moulds or bacteria.

A class of antibodies of particular interest is formed by

the Heavy Chain antibodies as found in Camelidae, like the
camel or the llama. The binding domains of these antibodies
consist of a single polypeptide fragment, namely the
variable region of the heavy chain polypeptide (HC-V). In
contrast, in the classic antibodies (murine, human, etc.),

the binding domain consists of two polypeptide chains (the
variable regions of the heavy chain (Vh) and the light chain
(V1)). Procedures to obtain heavy chain immunoglobulins from
Camelidae, or (functionalized) fragments thereof, have been
described in WO-A-94/04678 (Casterman and Hamers) and WO-A94/25591 (Unilever and Free University of Brussels).

Alternatively, binding domains can be obtained from the V_h fragments of classical antibodies by a procedure termed "camelization". Hereby the classical V_h fragment is

- transformed, by substitution of a number of amino acids, into a HC-V-like fragment, whereby its binding properties are retained. This procedure has been described by Riechmann et al. in a number of publications (J. Mol. Biol. (1996) 259, 957-969; Protein. Eng. (1996) 9, 531-537,
- Bio/Technology (1995) 13, 475-479). Also HC-V fragments can be produced through recombinant DNA technology in a number of microbial hosts (bacterial, yeast, mould), as described in WO-A-94/29457 (Unilever).
- Methods for producing fusion proteins that comprise an enzyme and an antibody or that comprise an enzyme and an antibody fragment are already known in the art. One approach is described by Neuberger and Rabbits (EP-A-194 276). A method for producing a fusion protein comprising an enzyme and an antibody fragment that was derived from an antibody

originating in Camelidae is described in WO-A-94/25591. A

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method for producing bispecific antibody fragments is described by Holliger et al. (1993) PNAS 90, 6444-6448.

A particularly attractive feature of antibody binding

behaviour is their reported ability to bind to a "family" of structurally related molecules. For example, in Gani et al.

(J. Steroid Biochem. Molec. Biol. 48, 277-282) an antibody is described that was raised against progesterone but also binds to the structurally-related steroids, pregnanedione, pregnanolone and 6-hydroxy-progesterone. Therefore, using the same approach, antibodies could be isolated that bind to a whole "family" of stain chromophores (such as the polyphenols, porphyrins, or caretenoids as described below). A broad action antibody such as this could be used to treat several different stains when coupled to a bleaching enzyme.

1.2.2. Peptides.

Peptides usually have lower binding affinities to the substances of interest than antibodies. Nevertheless, the binding properties of carefully selected or designed peptides can be sufficient to deliver the desired selectivity in an exidation process. A peptide which is capable of binding selectively to a fabric / ligand to which one would like to deliver a benefit agent, can for instance be obtained from a protein which is known to bind to that specific fabric / ligand. An example of such a peptide would be a binding region extracted from an antibody raised against that fabric / ligand. A suitable peptide could be analogous to the active center of a protein analogous to a non-catalytic binding domain of a protein, e.g. a receptor.

Alternatively, peptides that bind to such substance can be obtained by the use of peptide combinatorial libraries. Such a library may contain up to 10¹⁰ peptides, from which the peptide with the desired binding properties can be isolated. (R.A. Houghten, Trends in Genetics, Vol 9, no &, 235-239).

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Several embodiments have been described for this procedure (J. Scott et al., Science (1990) 249, 386-390; Fodor et al., Science (1991) 251, 767-773; K. Lam et al., Nature (1991) 354, 82-84; R.A. Houghten et al., Nature (1991) 354, 84-86).

Suitable peptides can be produced by organic synthesis, using for example the Merrifield procedure (Merrifield (1963) J.Am.Chem.Soc. 85, 2149-2154). Alternatively, the peptides can be produced by recombinant DNA technology in microbial hosts (yeast, moulds, bacteria) (K.N. Faber et al. (1996) Appl. Microbiol. Biotechnol. 45, 72-79).

1.2.3. Peptidomimics.

In order to improve the stability and/or binding properties of a peptide, the molecule can be modified by the incorporation of non-natural amino acids and/or non-natural chemical linkages between the amino acids. Such molecules are called peptidomimics (H.U. Saragovi et al. (1991) Bio/Technology 10, 773-778; S. Chen et al. (1992)

20 Proc.Natl.Acad. Sci. USA 89, 5872-5876). The production of such compounds is restricted to chemical synthesis.

1.2.4. Other organic molecules.

It can be readily envisaged that other molecular structures, which need not be related to proteins, peptides or derivatives thereof, can be found which bind selectively to fabrics / ligands to which one would like to deliver a benefit agent. For example, certain polymeric RNA molecules which have been shown to bind small synthetic dye molecules (A. Ellington et al. (1990) Nature 346, 818-922). Such binding compounds can be obtained by the combinatorial approach, as described for peptides (L.B. McGown et al. (1995), Analytical Chemistry, 663A-668A).

35 This approach can also be applied for purely organic compounds which are not polymeric. Combinatorial procedures

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for synthesis and selection for the desired binding properties have been described for such compounds (Weber et al. (1995) Angew.Chem.Int.Ed.Engl. 34, 2280-2282; G. Lowe (1995), Chemical Society Reviews 24, 309-317; L.A. Thompson et al. (1996) Chem. Rev. 96, 550-600). Once suitable binding compounds have been identified, they can be produced on a larger scale by means of organic synthesis.

A further embodiment would be for the domain with a high binding affinity to be a bispecific reagent. Such a reagent could fulfil the requirement of accumulating the benefit agent on the fabric either by supplying said reagent together with the benefit agent as a pre-formed non-covalent complex or by supplying the two separately and allowing them to self-assemble either in the wash liquor or on the fabric.

In the detergent composition according to the invention, the fusion protein can be used to deposit a Benefit Agent onto the fabric by means of the cellulose binding domain.

20 Alternatively, the domain having a high binding affinity for the fabric / ligand can be used to target the benefit agent.

In a further embodiment, the detergent compositions of the invention comprise micro-particles sensitised with CBD-antibody fusions, and configured such that the micro-particles are loaded with the benefit agent and the antibody domain of the fusion protein has a high affinity or specificity for a substance (or "marker molecule") typically found on some regions of fabrics but not on others. Examples of such marker molecules include bleach-damaged dyes and microbes known to be associated with malodour. The antibody domain targets the benefit agent to its intended site of action and binds it there. For example, Microbe-specific antibody based fusions may target fragrance-containing particles to the regions of malodour. Thus, a more efficient use of expensive ingredients is achieved. Alternatively,

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antibody based fusions specific for bleach-damaged dyes can target dyed particles to faded regions, thus replenishing the colour lost in the main wash cycle.

5 The CBD antibody fusion protein binds to the fabric via the CBD region, thereby allowing the antibody domain to bind to corresponding antigens / ligands that comprise or form part of the benefit agent. The fusion protein can be dosed in conjunction with the benefit agent or can be added to the fabric prior to the said fabric coming into contact with the benefit agent.

The benefit agent can be selected from softening agents, finishing agents/ protective agents, fragrances (perfumes), bleaching agents.

Examples of softening agents are clays, cationic surfactants or silicon compounds. Examples of finishing agents/
protective agents are polymeric lubricants, soil repelling agents, soil release agents, photo-protective agents (sunscreens), anti-static agents, dye-fixing agents, anti-bacterial agents and anti-fungal agents. The fragrances or perfumes may be encapsulated, e.g. in latex microcapsules. Suitable examples of bleaches are photobleaches. Examples of photobleaches are given in EP-A-379 312 (British Petroleum), which discloses a water-insoluble photobleach derived from anionically substituted porphine, and in EP-A-035 470 (Ciba Geigy), which discloses a textile treatment composition comprising a photobleaching component.

Another advantage of the present invention is that it is possible to target some benefit molecules to particular regions of fabric. For example, dyes can be targeted to colour-bleached regions to replenish dye lost in the main wash or fragrance can be targeted to regions where it is

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most needed (i.e. those regions where microbes associated with malodour are present - such as the "underarm" regions).

1.3 The fabrics

For laundry detergent applications, several classes of natural or man-made fabrics can be envisaged, in particular cotton. In the embodiment of the invention whereby the antibody region of the fusion protein targets the fabric, such macromolecular compounds have the advantage that they can have a more immunogenic nature, i.e. that it is easier to raise antibodies against them. Furthermore, they are more accessible at the surface of the fabric than for instance coloured substances in stains, which generally have a low molecular weight.

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An important embodiment of the invention is to use binding domains (as described above) that bind to several different types of fabrics. This would have the advantage of enabling a Benefit Agent to be deposited to several different types of fabric using the CBD-antibody fusion molecule.

2. The Detergent Composition.

The fusion proteins of the invention can be used in a detergent composition which is specifically suited for the purpose, and this constitutes a second aspect of the invention. When formulating a detergent composition, it is important to ensure that the other ingredients of the product are compatible with the activity of the fusion protein. WO-A-98/07820 (P&G) discloses inter alia rinse treatment compositions containing antibodies directed at cellulase and standard softener actives such as DEQA. The detergent product according to the present invention preferably contains no softener or low levels of softener active (e.g. HEQ).

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To that extent, the detergent composition comprises one or more benefit agents and optionally other conventional detergent ingredients. The invention in its second aspect provides a detergent composition which comprises from 0.1 - 50 % by weight, based on the total composition, of one or more surfactants. This surfactant system may in turn comprise 0 - 95 % by weight of one or more anionic surfactants and 5 - 100 % by weight of one or more nonionic surfactants. The surfactant system may additionally contain amphoteric or zwitterionic detergent compounds, but this in not normally desired owing to their relatively high cost. It may be be advantageous to also include cationic surfactants into the composition. Examples of suitable cationic surfactants are given in WO-A-97/03160 and WO-A-98/17767 (Procter&Gamble).

In general, the nonionic and anionic surfactants of the surfactant system may be chosen from the surfactants described "Surface Active Agents" Vol. 1, by Schwartz & Perry, Interscience 1949, Vol. 2 by Schwartz, Perry & Berch, Interscience 1958, in the current edition of "McCutcheon's Emulsifiers and Detergents" published by Manufacturing Confectioners Company or in "Tenside-Taschenbuch", H. Stache, 2nd Edn., Carl Hauser Verlag, 1981.

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Suitable nonionic detergent compounds which may be used include, in particular, the reaction products of compounds having a hydrophobic group and a reactive hydrogen atom, for example, aliphatic alcohols, acids, amides or alkyl phenols with alkylene oxides, especially ethylene oxide either alone or with propylene oxide. Specific nonionic detergent compounds are C_6 - C_{22} alkyl phenol-ethylene oxide condensates, generally 5 to 25 EO, i.e. 5 to 25 units of ethylene oxide per molecule, and the condensation products of aliphatic C_8 - C_{18} primary or secondary linear or branched alcohols with ethylene oxide, generally 5 to 40 EO.

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Suitable anionic detergent compounds which may be used are usually water-soluble alkali metal salts of organic sulphates and sulphonates having alkyl radicals containing from about 8 to about 22 carbon atoms, the term alkyl being 5 used to include the alkyl portion of higher acyl radicals. Examples of suitable synthetic anionic detergent compounds are sodium and potassium alkyl sulphates, especially those obtained by sulphating higher Ce-C18 alcohols, produced for 10 example from tallow or coconut oil, sodium and potassium alkyl C9-C20 benzene sulphonates, particularly sodium linear secondary alkyl C10-C15 benzene sulphonates; and sodium alkyl glyceryl ether sulphates, especially those ethers of the higher alcohols derived from tallow or coconut oil and 15 synthetic alcohols derived from petroleum. The preferred anionic detergent compounds are sodium C11-C15 alkyl benzene sulphonates and sodium C12-C18 alkyl sulphates. Also applicable are surfactants such as those described in EP-A-328 177 (Unilever), which show resistance to salting-out, 20 the alkyl polyglycoside surfactants described in EP-A-070 074, and alkyl monoglycosides.

Preferred surfactant systems are mixtures of anionic with nonionic detergent active materials, in particular the 25 groups and examples of anionic and nonionic surfactants pointed out in EP-A-346 995 (Unilever). Especially preferred is surfactant system which is a mixture of an alkali metal salt of a C16-C18 primary alcohol sulphate together with a C₁₂-C₁₅ primary alcohol 3-7 EO ethoxylate.

The nonionic detergent is preferably present in amounts greater than 10%, e.g. 25-90% by weight of the surfactant system. Anionic surfactants can be present for example in amounts in the range from about 5% to about 40% by weight of

35 the surfactant system.

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The detergent composition may take any suitable physical form, such as a powder, a tablet, an aqueous or non aqueous liquid, a paste or a gel. The fusion protein according to the invention will generally be used as a dilution in water of about 0.05 to 2%.

The detergent composition in accordance with the invention comprising the fusion protein can have any suitable form, i.e. the form of a granular composition, a liquid or a slurry of the enzyme, or with carrier material (e.g. as in EP-A-258 068 and the Savinase (TM) and Lipolase (TM) products of Novo Nordisk). A good way of adding the fusion protein to a liquid detergent product is in the form of a slurry containing from 0.005 to 50 % by weight of the complex in an ethoxylated alcohol nonionic surfactant.

The detergent compositions of the invention comprise about 0.001 to 50 mg, preferably from 0.01 to 10 mg of fusion protein per liter of the rinse liquor in use. A concentrated detergent composition before use will comprise about 1 to 1000 mg/l, preferably from 10 mg to 100 mg per liter of the detergent product.

The invention will now be further illustrated in the following, non-limiting Examples.

Example 1. CBD-anti Streptococcus sangius fusion (CBD-4715)

The ability of the CBD-fusion to capture bacteria onto a cellulosic matrix was investigated. The following protocol describes the method used to crosslink bacteria to cellulose via a CBD-antibody crosslinker, whereby the CBD region bound to the cellulose and the antibody domain binds to the bacteria: 1% phosphoric acid swollen cellulose was used to which 0-500µl 0.1 mg/ml CBD-4715 was added and volume made to 1ml with PBS pH 7. The samples were incubated for 30

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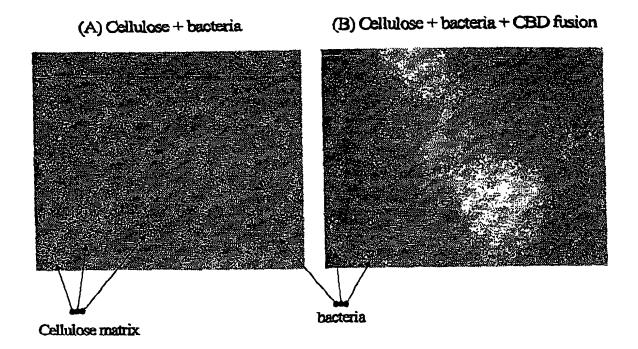
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minutes on a end over end shaker at 25°C. After each incubation step the samples were washed. To each tube 100µl 2.56×10⁶ Streptococcus sangius cells were added. Resulting samples were mixed for 10 minutes and then the samples were centrifuged and washed in PBS. An aliquot of each solution was removed and 10µl 0.1mg/ml carboxyfluorescein diacetate was added. This is a fluorescent marker for viable bacterial cells. Captured bacteria were visualised by green fluorescence (Excitation 488nm, Emission 525nm).



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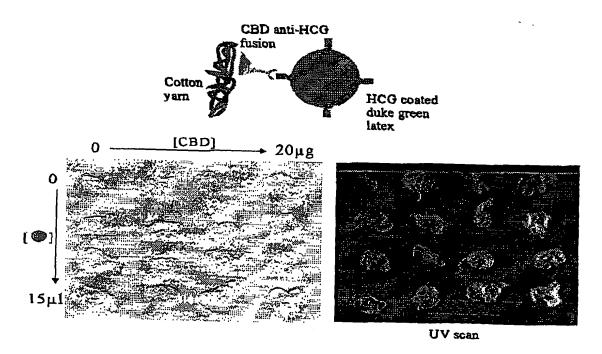
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Example 2. Latex capture on cellulose fibres (CBD-anti HCG fusion)

The ability of the CBD-anti HCG fusion to capture latex particles onto cotton yarns was tested: cotton yarns were washed thoroughly in 0.5ml 0.1M sodium carbonate pH 9.0. To these 0-12µl 0.4 mg/ml CBD-anti HCG fusion was added and incubated at 25°c for 10 minutes. The supernatants were removed and the yarns washed in 0.1M sodium carbonate pH 9.0. Yarns were then resuspended in 0.1M sodium carbonate pH 9.0 containing 0.25% Tween. 0-15µl duke green-HCG latex was added and samples were then incubated at 25°C for 20 minutes. Yarns were then washed with PBS-Tween and the yarns removed and dried. Latex remaining on the yarns was identified by fluorescence.

Single cotton yarns were used as the target surface. The incorporation of Tween was essential to remove the non-specific entrapment of latex on cotton. There was a good correlation between CBD fusion dose and degree of particle capture.



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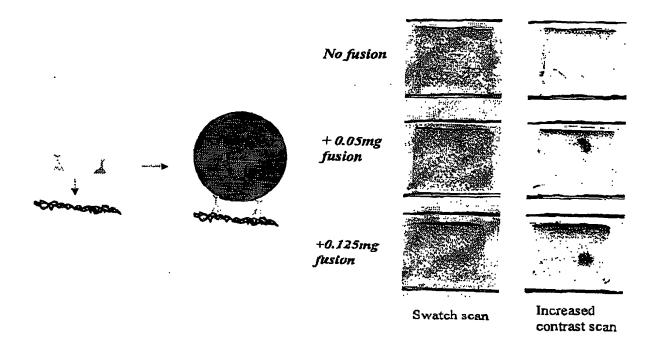
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Example 3.

Capture of coacervate particles containing a model benefit agent onto fabric via a CBD-antibody fusion protein crosslinker.

The next step was to investigate the ability to capture coacervate microcapsules onto cotton fabric using a CBD-fusion. The benefit agent in the capsule to be delivered was sunflower oil / ß-carotene sample as a model. Coacervate particles allow the incorporation of many hydrophobic and insoluble molecules. There was clear indication of enhanced deposition of particles to fabric, and the qualitative data produced suggested a dose response with the concentration of fusion protein. The untreated fabric did show entrapment of particles but there was a considerable increase in the delivery and deposition of capsules in the presence of the fusion protein.



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CLAIMS

- Fusion protein comprising a cellulose binding domain and a domain having a high binding affinity for another ligand.
- Fusion protein according to claim 1, wherein the cellulose binding domain is obtainable from a fungal enzyme origin such as Humicola, Trichoderma, Thermomonospora,
- Phanerochaete, Aspergillus or from a bacterial enzyme origin 10 such as Bacillus, Clostridium, Streptomyces, Cellulomonas and Pseudomonas.
- 3. Fusion protein according any one of the preceding 15 claims, wherein the cellulose binding domain is obtainable from Trichoderma reesei.
 - Fusion protein according to any one of the preceding claims, wherein the domain having a high binding affinity is an antibody or antibody fragment.
 - Fusion protein according to any one of the preceding claims, wherein the domain having a high binding affinity is a Heavy Chain antibody as found in Camelidae.
 - Fusion protein according to any one of the preceding claims, wherein the domain having a high binding affinity is a peptide.
- 30 Fusion protein according to any one of the preceding claims, wherein the domain having a high binding affinity is directed at a Benefit Agent.
- 8. Fusion protein according to claim 7, wherein the 35 Benefit Agent is selected from the group consisting of a fabric softening agents, fragrances, perfumes, polymeric

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lubricants, photoprotective agents, latexes, resins, dye fixative agents, encapsulated materials, antioxidants, insecticides, soil repelling agents or a soil release agents.

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- 9. Fusion protein according to claims 1-6, wherein the domain having a high binding affinity is directed at the fabric.
- 10 10. Fusion protein according to claim 9, wherein the fabric is polyester, or polyester / cotton, or wool.
- 11. Fusion protein according to claim 9, wherein the domain having a high binding affinity is directed at a specific15 part of the fabric.
- 12. Fusion protein according to any one of the preceding claims, wherein the cellulose binding domain is connected to the domain having a high binding affinity for another ligand by means of a linker consisting of 2-15, preferably 2-5 amino acids.
- 13. Fusion protein according to any one of the preceding claims, wherein the domain having a high binding affinity is directed at a micro-particles which are loaded with the benefit agent.
- 14. Fusion protein according to any one of the preceding claims, whereby the domain having a high binding affinity is a multi-specific antibody or antibody fragment or an analogous structure, whereby at least one specificity is directed to the fabric and the others are directed to one or more benefit agents.



15. Detergent composition comprising one or more surfactants and a fusion protein according to any one of the preceding claims.

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ABSTRACT

There is provided a fusion protein comprising a cellulose binding domain and a domain having a high binding affinity for another ligand. This high affinity binding domain is preferably directed at a Benefit Agent. The latter is preferably selected from the group consisting of a fabric softening agents, fragrances, perfumes, polymeric lubricants, photoprotective agents, latexes, resins, dye fixative agents, encapsulated materials, antioxidants, insecticides, soil repelling agents or a soil release agents. The fusion protein is advantageously used in a detergent composition to deliver the Benefit Agent to a fabric.

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